

## Note

### Synthesis and antimicrobial screening of 4*H*-2-benzoyl-3-hydroxy-3-methyl-2-phenyl 2,3-dihydro-furo[3,2-*c*]benzopyran-4-one and 4*H*-3-methyl-2-phenyl furo[3,2-*c*]benzopyran-4-one

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3-Acetyl-4-hydroxy-2*H*[1]benzopyran-2-one **2a-d** has been treated with bromodeoxybenzoin in NaOH, THF:HMPA to give 4*H*-2-benzoyl-3-hydroxy-3-methyl-2-phenyl 2,3-dihydro-furo[3,2-*c*] benzopyran-4-one **3a-d**. This on treatment with aqueous HCl in presence of dioxane gives 4*H*-3-methyl-2-phenyl furo[3,2-*c*] benzopyran-4-one **4a-d** through acid catalysed 1,2-elimination. All compounds have been screened for antimicrobial activity. They do not show significant activity.

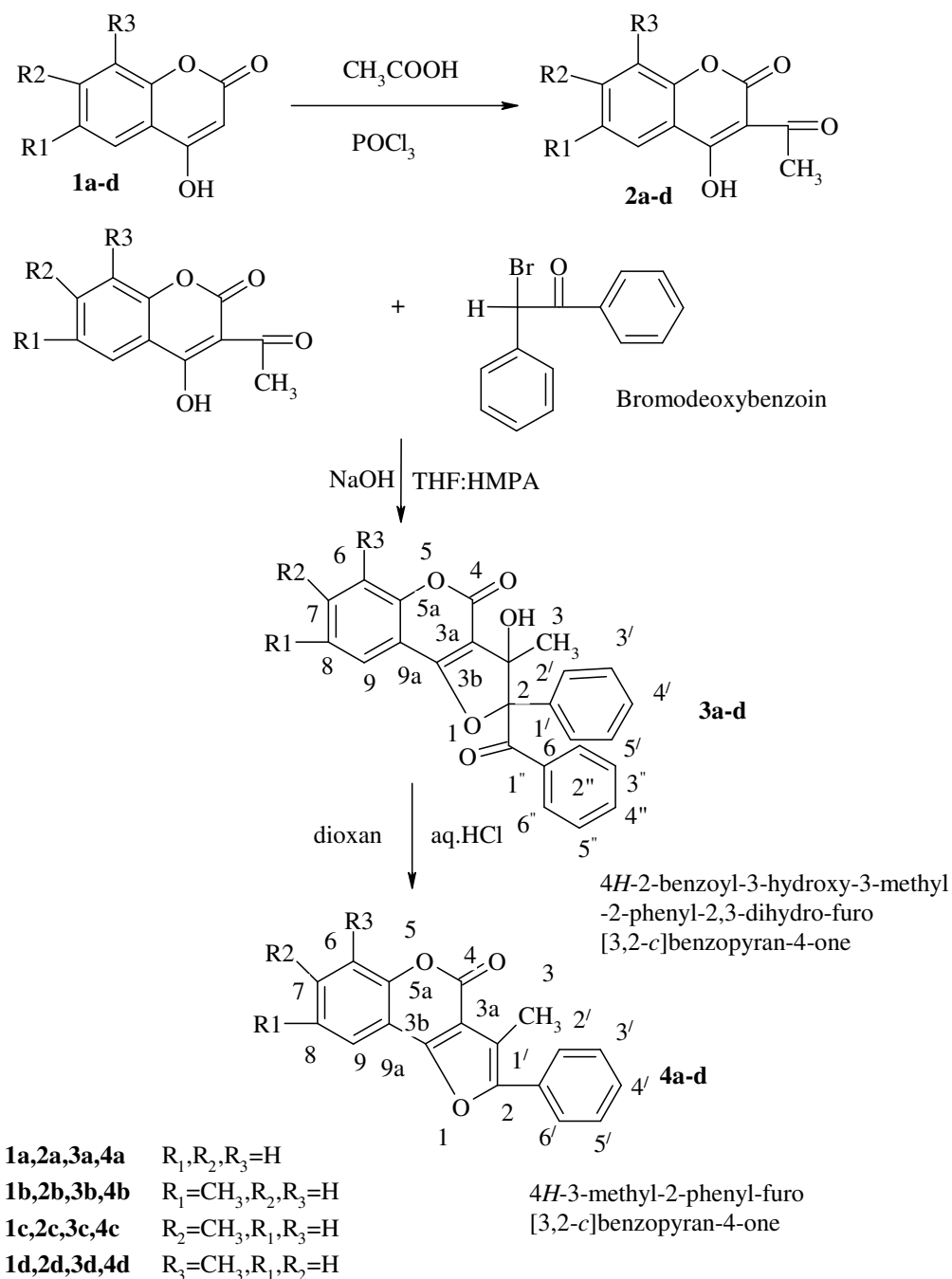
**Keywords:** Bromodeoxybenzoin, dioxane, furobenzopyran, herbicidal, antimicrobial activity

4-Hydroxycoumarin derivatives are of interest because of their anticoagulant<sup>1-7</sup>, spasmolytic<sup>8-10</sup>, rodenticidal<sup>11-14</sup> and estrogenic<sup>15</sup> activity. Some coumarin derivatives are known for their antifungal, antiinflammatory<sup>16-18</sup> and anti-HIV activity<sup>19-21</sup>. They are used as photosensitive drugs<sup>22</sup>, potent and selective human dopamine D<sub>4</sub> antagonists<sup>23</sup> and antibiotic agent<sup>24</sup>. Furobenzopyran is found in a variety of natural products exhibiting various types of herbicidal<sup>25</sup> activity. They have excellent herbicidal activity on weeds and are completely selective to food crops such as paddy rice and soyabeans as well as cotton. Furan derivatives constitute their own class of important drug and drug intermediates. It was therefore thought of synthesizing furan ring at fourth position of benzopyran moiety which possess some of the above mentioned biological activities.

3-Acetyl-4-hydroxy 2*H* [1] benzopyran-2-one<sup>26</sup> was treated with bromodeoxy- benzoin<sup>27</sup> using NaOH, THF:HMPA to give 4*H*-2-benzoyl-3-hydroxy-3-methyl-2-phenyl 2,3-dihydro-furo[3,2-*c*]benzopyran-4-one **3a-d** in good yields<sup>27</sup> (Scheme I). The IR

spectra of **3b** showed peak at 3404 cm<sup>-1</sup> for -OH stretching, 2924 cm<sup>-1</sup> for CH stretching, 1700 cm<sup>-1</sup> for carbonyl group of coumarin, 1694 cm<sup>-1</sup> for C=O stretching. The <sup>1</sup>H NMR of **3b** in DMSO-*d*<sub>6</sub> showed singlet at δ 1.22 for three protons of methyl group at C<sub>3</sub>, singlet at δ 2.25 for three protons of methyl group at C<sub>8</sub>, doublet at δ 7.75 (*J* = 7.5Hz) for a proton at C<sub>6</sub>, another doublet was observed at δ 8.0 (*J* = 7.5Hz) for a proton at C<sub>7</sub>. The singlet at δ 8.25 was assigned to the proton at C<sub>9</sub>. The hydroxyl proton appeared as a singlet at δ 5.25 which was D<sub>2</sub>O exchangeable. The multiplets between δ 7.0-8.25 accounted for rest of the ten protons. The <sup>13</sup>C NMR spectra displayed signals at δ 17 for methyl carbon of C<sub>3</sub>, δ 21 for methyl carbon of C<sub>8</sub>, signal appeared at δ 95 for C<sub>3</sub>, signal appeared at δ 101 for C<sub>3a</sub>, signal appeared at δ 116 for C<sub>9a</sub> and 120-140 for aromatic carbons, fused carbons at δ 152 for C<sub>5a</sub>. Signal appeared at δ 154 for C<sub>2</sub>, δ 167 for carbonyl carbon, signal appeared at δ 168 for C<sub>3b</sub>, δ 179 for carbonyl carbon attached to phenyl ring. The mass spectra gave molecular ion peak *m/z* (M<sup>+</sup>) at 412 (20%), 405, 393, 375, 281, 245, 203, 134, 91, 77 (100%), 65 along with other peaks.

4*H*-2-benzoyl-3-hydroxy-3-methyl-2-phenyl-2, 3-dihydro-furo[3,2-*c*] benzopyran 4-one **3a-d** on treatment with aqueous HCl in dioxane gave 4*H*-3-methyl-2-phenyl furo[3,2-*c*] benzopyran-4-one **4a-d** (Table I) through unusual acid catalysed 1,2-elimination<sup>28</sup> (Scheme I, Scheme II). The IR spectra of **4b** did not show peak at 3404 cm<sup>-1</sup> for -OH stretching. The IR spectra of **4b** showed peak at 2924 cm<sup>-1</sup> for CH stretching and 1700 cm<sup>-1</sup> for carbonyl group stretching. The <sup>1</sup>H NMR of **4b** in DMSO-*d*<sub>6</sub> showed singlet at δ 2.0 for three protons at C<sub>3</sub>, singlet at δ 2.25 for three protons of methyl group at C<sub>8</sub>, doublet at δ 7.76 (*J* = 7.5Hz) for a proton at C<sub>6</sub>, another doublet at δ 8.1 (*J* = 7.5Hz) for a proton at C<sub>7</sub>, singlet at δ 8.27 for proton at C<sub>9</sub> while rest of the five protons appeared as multiplets between δ 7.1-8.26. The <sup>13</sup>C NMR spectra displayed signals at δ 22 for methyl group at C<sub>3</sub>, δ 21 for methyl carbon at C<sub>8</sub>, δ 117 for C<sub>9a</sub>, signal appeared at δ 119 for C<sub>3a</sub>, δ 120-140 for aromatic carbons, fused carbons at δ 153 for C<sub>5a</sub>, signal appeared at δ 154 for C<sub>2</sub>, signal appeared at δ 159 for C<sub>3b</sub>, δ 167 for carbonyl carbon. The mass spectra gave molecular ion peak *m/z* (M<sup>+</sup>) at 290



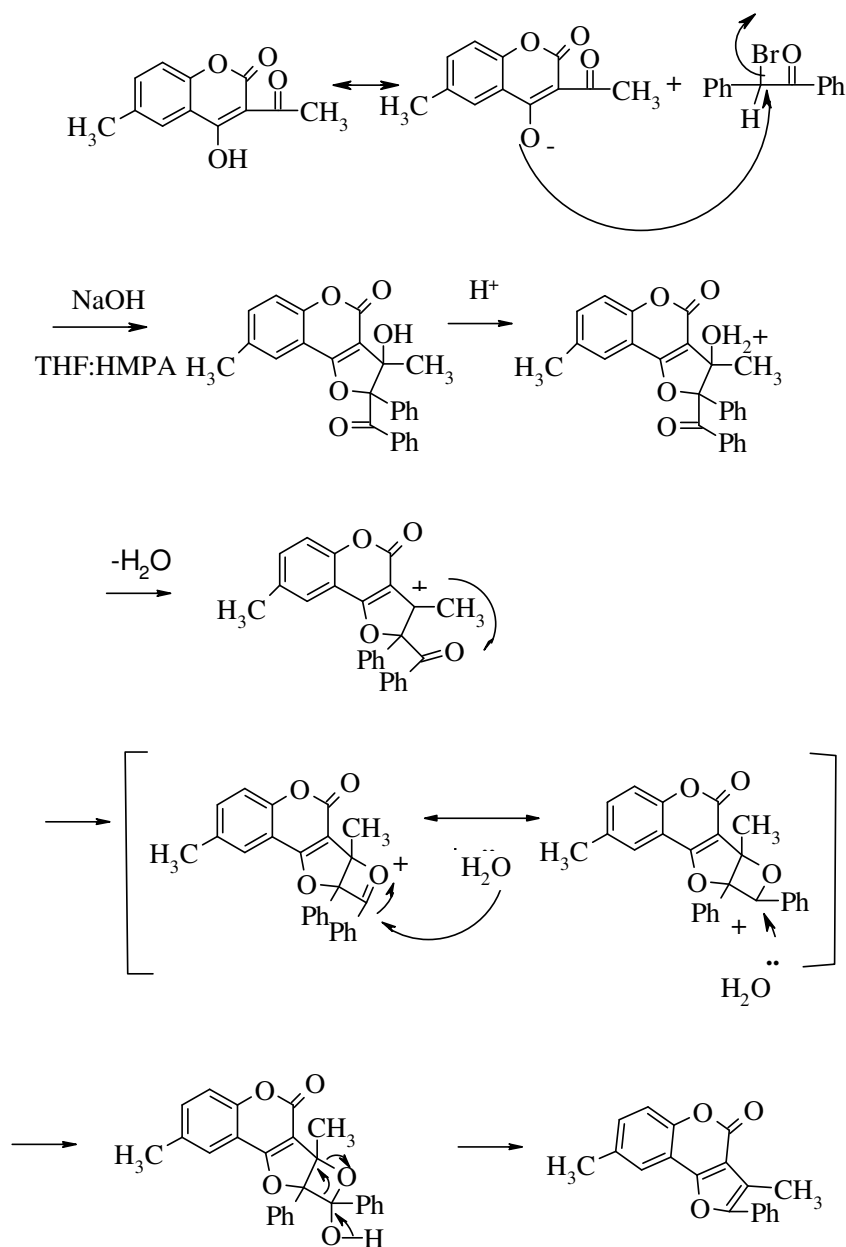
Scheme I

(45%), 279, 244, 228, 194, 133 (100%), 105, 104, 93, 78, 76 along with other peaks.

#### Antimicrobial activity

The above compounds **3a-d** and **4a-d** were screened for their antibacterial activity against *S. aureus*, *S. typhi* and *E. coli* and antifungal activity

against *A. niger* and *C. albicans*. The minimum inhibitory concentration (MIC) was determined using tube dilution method according to the standard procedure<sup>29</sup>. DMSO was used as a blank and Ciprofloxacin (MIC: 5 µg/mL) and Miconazole (MIC: 5 µg/mL) were used as antibacterial and antifungal standards respectively. An examination of the data



Scheme II

shows all the compounds had antibacterial activity ranging from 50 to 200  $\mu\text{g/mL}$  (Table II).

### Experimental Section

Melting points were recorded in open capillaries and are uncorrected. Homogeneity of the compounds

was checked on TLC. IR spectra ( $\nu_{\text{max}}$  in  $\text{cm}^{-1}$ ) were recorded on a Perkin-Elmer FT-IR instrument and  $^1\text{H}$  and  $^{13}\text{C}$  NMR on Jeol 300 MHz instrument using TMS as standard and mass spectra were recorded on a Shimadzu QP-2010 GC-MS. Biological testing were carried out at Padmaja Aerobiologicals (P) Ltd.

**Table I** — Characterization data of compounds **3a-d** and **4a-d**

Compd	Mol. formula	m.p. °C	Yield (%)	Found (%) Calcd		
				C	H	O
<b>3a</b>	C <sub>25</sub> H <sub>18</sub> O <sub>5</sub>	142	50	75.35 (75.37)	4.51 4.52	20.09 20.10)
<b>3b</b>	C <sub>26</sub> H <sub>20</sub> O <sub>5</sub>	145	60	75.71 (75.72)	4.84 4.85	19.40 19.41)
<b>3c</b>	C <sub>26</sub> H <sub>20</sub> O <sub>5</sub>	110	50	75.70 (75.72)	4.83 4.85	19.39 19.41)
<b>3d</b>	C <sub>26</sub> H <sub>20</sub> O <sub>5</sub>	120	50	75.69 (75.72)	4.82 4.85	19.38 19.41)
<b>4a</b> (Ref.30)	C <sub>18</sub> H <sub>12</sub> O <sub>3</sub>	150	60	78.25. (78.26	4.33 4.34	17.38 17.39)
<b>4b</b>	C <sub>19</sub> H <sub>14</sub> O <sub>3</sub>	122	65	78.61 (78.62	4.81 4.82	16.54 16.55)
<b>4c</b>	C <sub>19</sub> H <sub>14</sub> O <sub>3</sub>	100	65	78.60 (78.62	4.80 4.82	16.53 16.55)
<b>4d</b>	C <sub>19</sub> H <sub>14</sub> O <sub>3</sub>	117	60	78.59 (78.62	4.79 4.82	16.52 16.55)

**Table II** — Antibacterial activity of compounds **3a-d** and **4a-d**

Compd	Antibacterial activity µg/mL			Antifungal activity µg/mL	
	<i>E. coli</i>	<i>S. typhi</i>	<i>S. aureus</i>	<i>A. niger</i>	<i>C. albicans</i>
<b>3a</b>	160	200	80	50	60
<b>3b</b>	130	140	85	55	65
<b>3c</b>	150	150	90	80	90
<b>3d</b>	140	160	100	50	65
<b>4a</b>	160	200	110	90	100
<b>4b</b>	130	140	100	105	100
<b>4c</b>	150	150	115	95	100
<b>4d</b>	140	160	125	110	95
Ciprofloxacin(Std.)	5	5	5	-	-
Miconazole(Std.)	-	-	-	5	5

#### General procedure for the preparation of 4*H*-2-benzoyl-3-hydroxy-3-methyl-2-phenyl-2, 3-dihydro-furo[3,2-*c*]benzopyran-4-one, **3a-d**

3-Acetyl-2*H*-[1]-4-hydroxy-2-oxo-benzopyran **2a-d** (0.122 g, 1 mmol) was added to a stirred solution of NaOH (0.044 g, 1.1 mmol) and THF:HMPA(1:1, 4 mL) and the mixture was allowed to reflux for 30 min. To this, bromodeoxybenzoin (0.336 g, 1.1 mmole) in dry THF (2 mL) was added dropwise and the solution was refluxed for further 8 hr. On completion of the reaction, monitored by TLC, it was cooled and quenched with water. The reaction mixture was extracted with ether (3×50 mL) and washed with 5% KOH (3×10 mL), water (2×5mL)

and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Excess of solvent was removed on a rotary evaporator. The residual oil was then triturated with ethyl acetate and cooled to give a crystalline solid. It was filtered, dried and purified by recrystallization from hexane to give pure **3a-d**.

#### General procedure for the preparation of 4*H*-3-methyl-2-phenyl-furo[3,2-*c*] benzopyran-4-one, **4a-d**

To a solution of 4*H*-2-benzoyl-3-hydroxy-3-methyl-2-phenyl-furo[3,2-*c*]benzopyran-4-one, **3a-d** dissolved in dioxane (5 mL), aq. HCl (10%, 5 mL) was added and the mixture was refluxed for 10 hr.

The reaction mixture was quenched with water, extracted with ether, washed with water, dried over anhydrous  $\text{Na}_2\text{SO}_4$  and after removal of the excess solvent on a rotary evaporator, the residue was purified by recrystallization from hexane to give pure **4a-d**.

**3a:** IR(KBr): 3403 (-OH), 2923, 1710 ( $>\text{C}=\text{O}$ ), 1694, 1404, 873  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.21 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 5.24 (s, 1H, -OH,  $\text{D}_2\text{O}$  exchangeable), 7.1-8.25 (m, 14H, Ar-H).

**3b:** IR(KBr): 3404 (-OH), 2924, 1700 ( $>\text{C}=\text{O}$ ), 1693, 1400, 870  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.22 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 2.25 (s, 3H,  $\text{C}_8\text{-CH}_3$ ), 5.25 (s, 1H, -OH,  $\text{D}_2\text{O}$  exchangeable), 7.75 (d, 1H,  $\text{C}_6\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.0 (d, 1H,  $\text{C}_7\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.25 (s, 1H,  $\text{C}_9\text{-H}$ ), 7.0-8.25 (m, 10H, Ar-H);  $^{13}\text{C}$  NMR (DMSO- $d_6$ ):  $\delta$  17 ( $\text{C}_3\text{-CH}_3$ ), 21 ( $\text{C}_8\text{-CH}_3$ ), 95 ( $\text{C}_3$ ), 101 ( $\text{C}_{3a}$ ), 116 ( $\text{C}_{9a}$ ), 120-140 (aromatic carbons), 152 ( $\text{C}_{5a}$ ), 154 ( $\text{C}_2$ ), 167 ( $>\text{C}=\text{O}$ ), 168 ( $\text{C}_{3b}$ ), 179 ( $\text{O}=\text{C}-\text{C}_6\text{H}_6$ ); MS:  $m/z$  (%)  $\text{M}^+$  412 (20), 405, 393, 375, 281, 245, 203, 134, 91, 77 (100), 65.

**3c:** IR(KBr): 3402 (-OH), 2920, 1720 ( $>\text{C}=\text{O}$ ), 1690, 1401, 871  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.20 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 2.24 (s, 3H,  $\text{C}_7\text{-CH}_3$ ), 5.27 (s, 1H, -OH,  $\text{D}_2\text{O}$  exchangeable), 7.74 (d, 1H,  $\text{C}_8\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.0 (d, 1H,  $\text{C}_9\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.25 (s, 1H,  $\text{C}_6\text{-H}$ ), 7.1-8.24 (m, 10H, Ar-H).

**3d:** IR(KBr): 3409 (OH), 2937, 1730 ( $\text{C}=\text{O}$ ), 1690, 1457, 869  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.25 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 2.28 (s, 3H,  $\text{C}_8\text{-CH}_3$ ), 5.26 (s, 1H, -OH,  $\text{D}_2\text{O}$  exchangeable), 7.78 (d, 1H,  $\text{C}_7\text{-H}$ ,  $J=7.5$ ), 8.3 (t, 1H,  $\text{C}_8\text{-H}$ ), 8.26 (d, 1H,  $\text{C}_9\text{-H}$ ), 7.0-8.30 (m, 10H, Ar-H).

**4a:** IR(KBr): 1702 ( $>\text{C}=\text{O}$ ), 2925, 1403, 870  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.1 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 7.1-7.90 (m, 9H, Ar-H).

**4b:** IR(KBr): 1700 ( $>\text{C}=\text{O}$ ), 2924, 1401, 871  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.0 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 2.25 (s, 3H,  $\text{C}_8\text{-CH}_3$ ), 7.76 (d, 1H,  $\text{C}_6\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.1 (d, 1H,  $\text{C}_7\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.27 (s, 1H,  $\text{C}_9\text{-H}$ ), 7.1-8.26 (m, 5H, Ar-H);  $^{13}\text{C}$  NMR (DMSO- $d_6$ ):  $\delta$  22 ( $\text{C}_3\text{-CH}_3$ ), 21 ( $\text{C}_8\text{-CH}_3$ ), 117 ( $\text{C}_{9a}$ ), 119 ( $\text{C}_{3a}$ ), 120-140 (aromatic carbons), 153 ( $\text{C}_{5a}$ ), 154 ( $\text{C}_2$ ), 159 ( $\text{C}_{3b}$ ), 167 ( $>\text{C}=\text{O}$ ); MS:  $m/z$  (%)  $\text{M}^+$  290 (45), 279, 244, 228, 194, 133 (100), 105, 104, 93, 78, 76.

**4c:** IR(KBr): 1705 ( $>\text{C}=\text{O}$ ), 2926, 1406, 875  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.2 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 2.26 (s, 3H,  $\text{C}_7\text{-CH}_3$ ), 7.78 (d, 1H,  $\text{C}_8\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.1 (d, 1H,  $\text{C}_9\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.26 (s, 1H,  $\text{C}_6\text{-H}$ ), 7.0-8.25 (m, 5H, Ar-H).

**4d:** IR(KBr): 1708 ( $>\text{C}=\text{O}$ ), 2924, 1408, 879  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.1 (s, 3H,  $\text{C}_3\text{-CH}_3$ ), 2.29 (s, 3H,  $\text{C}_8\text{-CH}_3$ ), 7.77 (d, 1H,  $\text{C}_7\text{-H}$ ,  $J=7.5\text{Hz}$ ), 8.2 (t, 1H,  $\text{C}_8\text{-H}$ ), 8.25 (d, 1H,  $\text{C}_9\text{-H}$ ,  $J=7.5\text{Hz}$ ), 7.2-8.25 (m, 5H, Ar-H).

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## References

- 1 Nagashima R, Reilly R O' & Levy G, *Clin Pharm Ther*, 1, **1969**, 22.
- 2 Reilly R O', Ohms J & Motley C, *J Biol Chem*, 244, **1969**, 1303.
- 3 Numamoto N, Yamamoto T, Yamashita A, Hatano T & Kurahara J, *Jpn Kokai Tokkyo Koho, JP*, **1990**, 02 36 178 [90 30 178].
- 4 Manolov I & Danchev N D, *Eur J Med Chem*, 30, **1995**, 531.
- 5 Wallin R & Hutson S M, *Trends Mol Med*, 10, **2004**, 299.
- 6 Trivedi K N, Madhava Rao S S, Mistry S V & Desai S M, *J Indian Chem Soc*, 78, **2001**, 579.
- 7 Smirnova T V, Vishnyakova G M, Lakin K M, Novikova N V, Lobanova E G, Zonova G N & Sklyarenko V I, *Tr Inst Mosk, Khim Tekhnol Inst im D I Mendeleeva*, 149, **1987**, 92.
- 8 Boschetti E, Molho D & Fontaine L, *S African Patent*, **1968**, 68 01, 383,02.
- 9 Stacchino C, Spano R & Pettite A, *Bull Chim Farm*, 122, **1983**, 158.
- 10 Kostova, Manolov, Nicolov, Konstantinov & Karaivanova, *Eur J Med Chem*, 36, **2001**, 339.
- 11 LIPHA (Lyonnaise Industrielle Pharmaceutique), *GB Patent*, **1968**, 1, 252, 088.
- 12 McIntyre J S & Knight A R, *US Patent*, **1970**, 3 511 856.
- 13 Szejtli J, Szente L, Tarak S, Voroshazi L, Harshegyi J, Daroczi I, Zajak A & Teljes, *Hungarian Patent*, **1984**, 38, 203.
- 14 Whittle A J, *Eur Pat Appl, EP*, **1986**, 175 466.
- 15 Mazzei, Dondero, Sottofattori, Melboni & Minafra, *Eur J Med Chem*, 36, **2001**, 851.
- 16 Jacquot, Bermont, Giorgi, Refoulet, Adessi, Daubrosse & Xicluna, *Eur J Med Chem*, 36, **2001**, 127.
- 17 Mazzei M, Garzoglio R, Sottofattori E, Melloni E & Michetti M, *Il Farmaco*, 52, **1999**, 539.
- 18 Mazzei M, Sottofattori E, Dondero R, Ibrahim M, Melloni E & Michetti M, *Il Farmaco*, 54, **1999**, 452.
- 19 Zhao H, Neamati N, Hang H, Mazumder A, Wang S, Sunder S, Milne G W A, Pommier Y & Burke Jr T R, *J Med Chem*, 40, **1997**, 242.
- 20 Marcal de QUEIROZ P Italy *PCT Int Appl WO*, **1998**, 98 25,608.
- 21 Skulnik H I, Johnson P D, Aristoff P A, Morris J K & Lovasz K D, *J Med Chem*, 40, **1997**, 1149.
- 22 Wulf H, Rauer H, Düring T, Hanselmann C, Ruff K, Wirish A, Grissmer S & Hänsel W, *J Med Chem*, 41, **1998**, 4542.

- 23 Kesten S R, Heffner T G, Johnson S J, Pugley T A, Wright J L & Wise L D, *J Med Chem*, 42, **1999**, 3718.
- 24 Crow F W, Duholke W K, Farley K A, Hadden C E, Hahn D A, Kahizny B D, Mallory C S, Martin G E, Smith R F & Thamann T J, *J Heterocyclic Chem*, 36, **1999**, 365.
- 25 Arai, Ooka, Koizumi, Koda, Iwasaki, Kanemoto & Mobara, *US Patent*, Japan, **1994**, 5, 356, 866.
- 26 Klose J, *Arch Pharm*, 288, **1955**, 356.
- 27 Crenshaw R R, Jeffries A T, Luke G M, Cheney L C & Bialy G, *J Med Chem*, 14, **1971**, 1185.
- 28 Jha, Sharma, Maulik, Yadav & Hajela K, *Indian J Chem*, 43B, **2004**, 1341.
- 29 Frankel S, Reitman & Sonnenwirth A C, *Gradwol's Clinical Laboratory Methods and Diagnosis*, A textbook on laboratory procedure and their interpretation, Vol. 2, 7<sup>th</sup> edn, (C V Mosby Company, Germany), **1970**, 1406.
- 30 Francesco R, Giovanni G, Francesco F & Cristina B, *Tetrahedron Lett*, 42, **2001**, 3503.